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c) subliming / sublimation. d) subliming / sublimation 2 a) solid liquid gas Note: Solids should have regularly packed particles touching. Liquids should have most of the particles touching at least some of their neighbours, but with gaps here and there, and no regularity.

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(c) The atomic radii decreases across Period 3 (1) as the number of protons in the nucleus increases so does the nuclear charge (1). This results in an increase in the attractive force between the nucleus and the outer electrons.

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